

I. COURSE DESCRIPTION:

This is an introductory course to provide Fish & Wildlife and Parks and Outdoor Recreation Technician students with a fundamental understanding of chemical interaction in nature. Three main topics will be examined, including: chemical characteristics of water in relation to life, systems in nature, and chemical pollutants.

II. LEARNING OUTCOMES AND ELEMENTS OF THE PERFORMANCE:

Upon successful completion of this course, the student will demonstrate the ability to:

1. **Explain various characteristics of water as they relate to life systems.**

Potential Elements of the Performance:

- describe various chemical and physical properties of water including: density, viscosity, polarity, surface tension, specific heat, solubility and pH
- determine the dissolved oxygen and carbon dioxide concentration of water
- determine the alkalinity of water
- separate materials on the basis of their physical properties
- explain thermal stratification of lakes and lake turnover

2. **Demonstrate the relationships of chemical interactions to important global environmental issues.**

Potential Elements of the Performance:

- explain bio-magnification
- describe the causes for and impacts of the following global environmental problems:
 - i) acid deposition
 - ii) ozone depletion
 - iii) global warming
 - iv) eutrophication
 - v) toxic pollution

3. Explain the systems approach to analyzing natural processes.

Potential Elements of the Performance:

- identify important aspects of a system, including: components, state and couplings
- explain and give examples of feedback loops
- explain what is meant by equilibrium
- demonstrate how the systems approach is applied in identifying and resolving environmental issues

4. Develop safe and correct lab technique with chemical handling and instrument use.

Potential Elements of the Performance:

- discuss and apply safe lab procedures including handling of dangerous chemicals
- demonstrate knowledge of the Workplace Hazardous Materials Information System
- demonstrate use of balances, pH metre, glassware including pipettes and burettes, hot plates, fume hoods
- prepare solutions
- conduct titration procedures

III. TOPICS:

1. **Characteristics of Water**
2. **The Systems Approach**
3. **Chemical Pollutants**

IV. REQUIRED RESOURCES/TEXTS/MATERIALS:

Science & Nature 2 Lab Manual, lab coat

V. EVALUATION PROCESS/GRADING SYSTEM:

1 Unit Test	20%
1 Final Exam	30%
<u>3 Labs</u>	<u>50%</u>
TOTAL	100%

Late lab assignments and report values will be reduced at a rate of 10% per day for a maximum period of 5 days after the due date. After 5 days the lab assignment/report value will be zero.

All lab assignments and reports must be submitted regardless of lateness to pass the course.

Failure to attend a test without medical or severe personal reasons will result in a zero and no opportunity to make up the test will be offered.

Course re-writes are not offered.

The following semester grades will be assigned to students in postsecondary courses:

<u>Grade</u>	<u>Definition</u>	<u>Grade Point Equivalent</u>
A+	90 - 100%	4.00
A	80 - 89%	3.75
B	70 - 79%	3.00
C	60 - 69%	2.00
R (Repeat)	59% or below	0.00
CR (Credit)	Credit for diploma requirements has been awarded.	
S	Satisfactory achievement in field placement or non-graded subject areas.	
U	Unsatisfactory achievement in field placement or non-graded subject areas.	
X	A temporary grade. This is used in limited situations with extenuating circumstances giving a student additional time to complete the requirements for a course (see <i>Policies & Procedures Manual – Deferred Grades and Make-up</i>).	
NR	Grade not reported to Registrar's office. This is used to facilitate transcript preparation when, for extenuating circumstances, it has not been possible for the faculty member to report grades.	

VI. SPECIAL NOTES:

Special Needs:

If you are a student with special needs (e.g. physical limitations, visual impairments, hearing impairments, or learning disabilities), you are encouraged to discuss required accommodations with your instructor and/or the Special Needs office. Visit Room E1204 or call Extension 493, 717, or 491 so that support services can be arranged for you.

Retention of course outlines:

It is the responsibility of the student to retain all course outlines for possible future use in acquiring advanced standing at other postsecondary institutions.

Plagiarism:

Students should refer to the definition of “academic dishonesty” in *Student Rights and Responsibilities*. Students who engage in “academic dishonesty” will receive an automatic failure for that submission and/or such other penalty, up to and including expulsion from the course/program, as may be decided by the professor/dean. In order to protect students from inadvertent plagiarism, to protect the copyright of the material referenced, and to credit the author of the material, it is the policy of the department to employ a documentation format for referencing source material.

Course outline amendments:

The Professor reserves the right to change the information contained in this course outline depending on the needs of the learner and the availability of resources.

Substitute course information is available in the Registrar's office.

VII. PRIOR LEARNING ASSESSMENT:

Students who wish to apply for advanced credit in the course should consult the professor. Credit for prior learning will be given upon successful completion of a challenge exam or portfolio.

VIII. DIRECT CREDIT TRANSFERS:

Students who wish to apply for direct credit transfer (advanced standing) should obtain a direct credit transfer form from the Dean's secretary. Students will be required to provide a transcript and course outline related to the course in question.